

Kandi Raj College
B.Sc. 3rd Semester Hons. Internal Assessment examination
Subject: Physical Chemistry [CHEMHT-5]

Time:

F.M. 10

Answer any two

1. (a) Justify or criticise the following statement: $n=0$ is not possible for a particle in a box.

(b) Calculate the linear momentum of photons of wavelength 350 nm. What speed does a hydrogen molecule need to travel to have the same linear momentum? 2+3=5

2. (a) Write Short notes on exergonic and endergonic reactions
(b) Write down the Gibbs-Duhem equation explaining the terms. What is the significance of Gibbs-Duhem equation?
2+1+2=5

3. (a) What is the dimension of viscosity? Give two examples of high viscosity liquids and two examples of low viscosity liquids.
(b) In an X-ray photoelectron experiment, a photon of wavelength 121 pm ejects an electron from the inner shell of an atom and it emerges with a speed of 56.9 m s^{-1} . Calculate the binding energy of the electron.
3+2=5

4. (a) Define eigenfunction.
(b) Identify which of the following functions are eigenfunctions of the operator d/dx
i) e^{kx}
ii) $\cos kx$
iii) kx
iv) e^{-ax}
Give the corresponding eigenvalue where appropriate.

1+4=5

Kandi Raj College
Internal Assesment-2020
Class- B.Sc(Hons) 3rd Sem
Sub. Inorganic Chemistry
Course Code: CEMH/CC-T-06

Full Mark-10

Time-

Answer any five questions

1. Explain the bond angle of the following mole?
 OF_2 , OCl_2 , OH_2
2. Explain the state of hybridization of $\text{Ni}(\text{CO})_4$ and calculate its magnetic moment?
3. Compare the hydration enthalpy of K^+ and F^- ?
4. $r_{\text{Ca}^{2+}}$ is 113pm and r_{F^-} is 135 pm, predict the crystal geometry?
5. MgCO_3 thermally less stable than CaCO_3 -explain?
6. Solubility of the AgX in the H_2O decreases from fluoride to iodide- Explain?
7. Explain why PbCl_2 is white while PbI_2 is yellow?

Choose the right option —

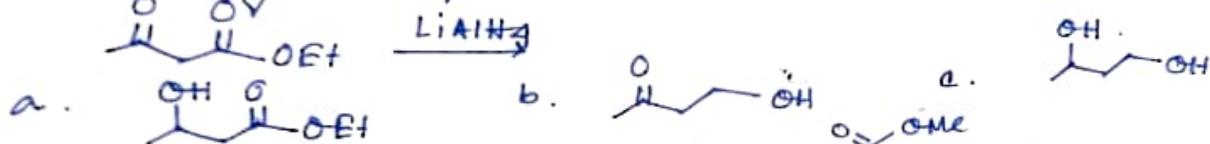
5 x 2 = 10

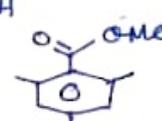
1. 1°-kinetic isotopic eff. is observed for —
 a. chlorination of C_4H_6 b. nitration of C_4H_6 c. sulfonation of C_4H_6 .

2. $\text{PhCH}=\text{CH}_2 \longrightarrow \text{PhCH}_2\text{CH}_2\text{OH}$, for this hydration suggest proper reagent —

- a. H_3O^+ b. $\text{Hg}(\text{OAc})_2, \text{NaBH}_4$ c. $\text{SiO}_2/\text{OH}^-, \text{H}_2\text{O}_2/\text{OH}^-$

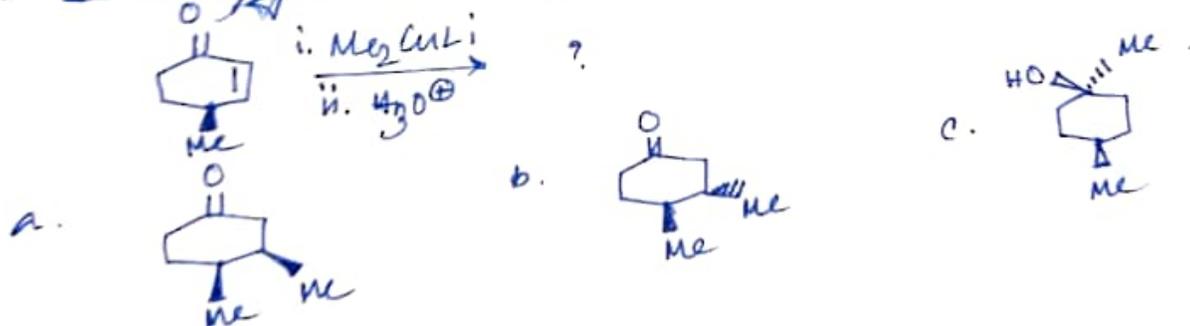
3. Identify the prod —



4. Acid catalysed hydrolysis of  follows —

- a. AAc^1 b. AAc^2 c. AAc^2

5. Identify the major prod —



Kandi Raj College
Internal Assesment-2020
Class- B.Sc(Hons) 3rd Sem
Sub. Inorganic Chemistry
Course Code: CEMH/SEC-1

Full Mark-10

Time-

Answer any five questions

1. Draw the structure of $Zn(EDTA)^{2-}$ complexe.?
2. What is BOD?
3. Give an example of two food preservating agent ?
4. Write the difference between TLC and paper chromatography?
5. Draw the structure of the indicator which is used in complexometry titrations ?
6. Write the two protocols for sampling?

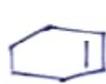
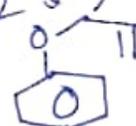
Kandi Raj College
Dept. of Chemistry

Paper - CHEM - HGE - T - 3

নিম্নলিখিত প্রশ্নগুলির উত্তর দাও

5 x 2 = 10

1. কার্নো-চক্র (Carnot cycle) কতটি কাজ করে?
2. ধাতুর অক্সিডেশন কোথায় ঘটে? অক্সিডেশন কোথায় ঘটে? উল্লেখ করো।
3.
$$\text{NH}_3(g) + \text{HCl}(g) \rightleftharpoons \text{NH}_4\text{Cl}(s) + \text{heat}$$

এই বিক্রিয়ায় সমতা স্থাপন করা যায় কীভাবে? উল্লেখ করো।
4. বিক্রিয়ায় সমতা স্থাপন করার জন্য কীভাবে কাজ করবে? উল্লেখ করো।
5. নিম্নলিখিত বিক্রিয়ায় বিক্রিয়ক ও উৎপাদন লিখো।
 - a. $\text{PhCHO} \xrightarrow[\text{50\%}]{\text{NaOH}}$?
 - b. $\text{Me}_2\text{C}(\text{OH})-\text{C}(\text{OH})\text{Me}_2 \xrightarrow{\text{H}^+}$?
 - c.  $\xrightarrow[\text{KMnO}_4]{\text{alkaline}}$?
 - d.  $\xrightarrow{\Delta}$?