661/4 Chem/PR UG/5th Sem/Chem(H)DSE-2B/PR/20

U.G. 5th Semester Examination - 2020

CHEMISTRY

[HONOURS] Discipline Specific Elective (DSE) Course Code : CHEM(H)DSE-P-2B/PR [PRACTICAL]

Full Marks : 20 Time : 2 Hours The figures in the right-hand margin indicate marks.

Answer any **four** questions:

 $5 \times 4 = 20$

- 1. a) Explain pink coloured $[Co(H_2O)_6]^{2+}$ turns into blue $[Co(H_2O)_4]^{2+}$ on dehydration.
 - b) Why hydrogen bonded O-H appears as broad band?
 - c) Which compound is used as internal standard during D_2O exchange? 2+2+1=5
- a) Write down the some safety precaution may be useful for minimization of accident in laboratory.
 - b) What is complexometric titration?
 - c) How will you determine metallic element in food and beverage industry? 2+1+2=5

[Turn over]

- 3. a) What is the basic principle of IR spectra?
 - b) Show the schematic diagram of Gas Chromatograph. 2+3=5
- 4. a) How does HPLC help in speciation of organic pollutant in environmental sample?
 - b) Write down the function of Glass electrode. 3+2=5
- 5. a) How can you separate carbohydrates HPLC?
 - b) Write down the types of liquid-liquid extraction may exist?
 - c) How can you check purity of sample of phenol?
 2+2+1=5
- 6. a) Why only KCl and NH_4NO_3 are used in salt bridge?
 - b) Why voltameters are not used for EMF measurement?
 - c) What is the function of Nicolprism?

(2)

2+2+1=5

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