586/1 Chem.

UG/5th Sem/CHEM-H-DSE-T-2A/21

U.G. 5th Semester Examination - 2021

CHEMISTRY

[HONOURS]

Discipline Specific Elective (DSE)
Course Code: CHEM-H-DSE-T-2A

Full Marks : 40 Time : $2\frac{1}{2}$ Hours

The figures in the right-hand margin indicate marks.

Candidates are required to give their answers in their own words as far as practicable.

1. Answer any **five** questions:

 $2 \times 5 = 10$

- a) What do you mean by standard deviation of a population of observations?
- b) What is t-test?
- c) Define distribution coefficient.
- d) How will you define exchange capacity of a cation exchange resin?
- e) Classify electroanalytical method.
- f) State Lambert-Beers law.
- g) What is interferogram?
- h) Define dark current.
- 2 Answer any **two** questions:

 $5 \times 2 = 10$

a) Explain basic principle of thermogravimetry. 5

[Turn over]

- b) What is the difference between precision and error? What is random error? 3+2=5
- c) Write a short note on role of computer in instrumental method of analysis. 5
- d) Write a short note on ion chromatography. 5
- 3. Answer any **two** questions:

 $10 \times 2 = 20$

- a) Explain the factors that control solvent extraction. What is synergic extraction? Explain with example. What do you mean by size exclusion chromatography? 4+3+3=10
- b) Explain the instrumentation of HPLC using a simple schematic diagram. What are the differences between adsorption chromatography and partition chromatography?

 What is suppressor column? 4+3+3=10
- c) What are the differences between single beam and double beam UV-VIS? What are the advantages of FT-IR? Explain the advantages of F-test in analytical measurement. 3+3+4=10
- d) How will you determine total iron spectrophotometrically? Write a note on effect and importance of isotope substitution.

 Describe the principle of iodometric determination of Cu²⁺.

 3+4+3=10