Kandi Raj College Internal Examination-2022 2nd Semester, Chemistry Hons, Paper CC – 3

Answer any 5 questions

- 1. Prove that $C_P C_v = \alpha^2 TV/\beta$, $\alpha = Coefficient of Thermal Expansion, \beta = Coefficient of Isothermal Compressibility$
- 2. State and prove Clausius Inequality.
- 3. Rate constant of a reaction is found to be doubled when temperature is raised from 27 $^{\circ}$ C to 37 $^{\circ}$ C. Find out Activation Energy.
- Show in Lineweaver Burk Plot in enzyme catalysis Slope/Intercept = K_M, Michaelis Constant.
- Calculate the equilibrium constant for the redox reaction between Sn(s) and Pb(II) ion. Given E^o_{red} (Pb²⁺/Pb)=-0.126v and E^o_{red} (Sn²⁺/Sn)=-0.126v.
- 6. Calculate the working potential of diphenyl ammine indicator. Given $E^{o}_{red}(In_{ox}/In_{red}){=}0.77v$
- 7. What is Disproportionation reaction. Give an example.
- 8. Why addition of Phosphoric acid is essential in the titration of Fe(II) with dichromate solution in presence of BDS indicator.

5x2=10

KANDI RAJ COLLEGE

U.G. 2nd Semester Internal Examination-2022

CHEMISTRY HONOURS

Paper: CHEMHT-4

marks: 1

Time: 1 hour

1) Answer any five questions from the following: 2x5=10

i) Explain the nature of stereoisomerism exhibited by the compounds of the formula $abC=(C=)_nCab$ where n=1 and 2.

ii) Write the stereostructure of the alcohol obtained by the attack of hydride on 2-butanone from its Si-face.

iii) Draw the conformational energy diagram for n-butane for rotation around C₂-C₃ bond. Show all the conformations.

iv)What is activation energy of a chemical reaction? Draw the energy profile diagram for a single step reaction.

v) What is meant by primary kinetic isotope effect? Give Example.

vi) Guanidine (H2N-C(=NH)-NH2) is a very strong base. Explain.

vii) Draw the energy profile diagram of S_N1 reaction considering a suitable example.

viii) Benzyl chloride (PhCH₂Cl) is more reactive than Ethyl chloride (CH₃CH₂Cl) both in S_N1 and S_N2 conditions. Explain.

ix) Between CH₃-CH₂-CH₂-Cl and CH₃-O-CH₂-Cl, which would react faster in S_N1 reaction? Explain.

x) E2 and E1cB reactions are kinetically indistinguishable. Explain.

